

## APPENDIX

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APPENDIX (I)

PRE-TEST

Time : 2 Hrs.

Marks : 50

- N.B. : 1) All questions are compulsory  
2) Write answers of the questions in the space provided below the question.

1. Draw the structure of diamond & Graphite. (4)

Diamond

Graphite

2. Give three points of differences between diamond and Graphite. (3)

Diamond

Graphite

1)

2)

3)

3. List the important sources of energy of in our life. (2)

(i)

(iii)

(ii)

(iv)

4. Write the meaning of :

(a) Renewable sources of energy : (2)

(b) Non-Renewable sources of energy : (2)

5. What is an atom ? Draw it's structure. (4)

6. Give structural and electron dot formula of the following : (6)

(i) Methane.  
Structural  
Formula :

(ii) Butene  
Structural  
Formula :

Electron dot  
Formula :

Electron dot  
Formula :

(iii) Pentyne  
Structural  
Formula :

(iv) Ethene  
Structural  
Formula :

Electron dot  
Formula :

Electron dot  
Formula :

7. Give molecular formula for the following compounds after giving general formula. (9)

(1) Alkane :  
General Formula :

Formula For  $C_{15}$  :

Formula For  $C_{20}$  :

(2) Alkene :  
General Formula :

Formula For  $C_{25}$  :

Formula For  $C_{10}$  :

(3) Alkyne :  
General Formula :

Formula For  $C_{39}$  :

Formula For  $C^{18}$  :

8. Define Isomerism ? Give isomers of butane and pentane. (8)

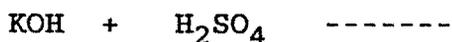
Definition of isomerism :

Isomers of Butane :

Isomers of Pentane :

9. Write Chemical Equation showing the neutralization of caustic potash (KOH) with sulphuric acid. Define the process of neutralization. (4)

Chemical Equation :



Definition of Neutralization process :

10. Give scientific reasons :

(a) Yellow phosphorous is kept under water. (3)

(b) Non-metallic elements are different from the metallic elements. (3)