LIST OF SYMBOLS AND ABBRIVIATIONS

~

,

~

•	-	-
a	-	Standard hardness
a ₁	-	intercept on log d axis in the plot of log p vs. log d of LLR
a 2	-	intercept on log d axis in the plot of log p vs. log d of HLR
a, b, c	-	lattice parameters of unit cell
A	-	unit area
b		constant(depends upon material)
c, cocn	-	concentration of solution
С	-	constant depending upon the indenter shape
đ	-	diagonal length of indentation mark
đ.	-	diagonal length of indentation mark
	-	ionic diameter short hand notation used for $d_{h_1k_1l_1}$
đ	-	ionic diameter
đ	-	ionic diameter short hand notation used for $d_{h_1k_1l_1}$
d d ₁ , d ₂	-	ionic diameter short hand notation used for $d_{h_1k_1l_1}$ and $d_{h_2k_2l_2}$
d d ₁ .d ₂ d-AHT	-	ionic diameter short hand notation used for $d_{h_1k_1l_1}$ and $d_{h_2k_2l_2}$ ammonium hydrogen d-tartrate crystal

•

: S-2 :

- dva. dva average of two diagonal lengths of vickers indentation mark
- d_{v_X} , dv_X diagonal length of vickers indentation mark measured along direction [001] on the cleavage plane and on m-face and along direction [101] on z-face
- d_{vy}, dvy diagonal length of vickers indentation mark measured normal to direction [001] on cleavage plane and on m-face and normal to [101] on z-face
- d(001) diagonal length of vickers indentation mark measured along direction [001] on cleavage plane
- d(100) diagonal length of vickers indentation mark measured along direction [100] on the cleavage plane
- D distance
- D diffusion constant
- E modulus of elasticity (young's modulus)
- E extra-ordinary ray
- E_C energy required for formation of crystal nucleus

:	5-3	:
÷	తాన	- 7

		÷ -3−-3 ÷
^E s .	-	activation energy for surface dissolution
^E t	•••	activation energy for tangential dissolution
FS	-	feed solution (ammonium chloride solution)
G	-	free energy of two dimentional nucleus
G	-	Galelian telescope
h	-	planck's constant
н	-	half-shade device (Laurent's plate)
Ĥ	-	average hardness number in HLR
H _k	-	knoop hardness number
H _k	-	average knoop hardness number in HIR
HLR	-	high load region
H _v	-	average vickers hardness number in HLR
H _{va} , H _{va}	-	vickers hardness number considering average of two diagonal lengths of vickers indentation mark in account
H _{vx} , H _{vx}	-	vickers hardness number along direction $[001]$ on the cleavage and on m-face and along direction $[101]$ on z-face
H _{vy} , Hvy	-	vickers hardness number normal to direction [001] on the cleavage plane and on m-face and along [101] on z-face

			: S=}+ :
	k	-	Boltzmannconstant
	к	-	constant
	Kw	-	ionic product of water
	1	-	length of tube of polarimeter
	L	-	convex lens of polarimeter
	L	•••	the heat of dissolution
	10	-	least count
	LLR		low load region
	m	-	slope of straight line plot
	m	-	gram of optical active substance in 100 c.c. of distilled water
	m	-	slope of the plot of log $\overline{H}\Gamma_Q$ vs. log T_Q
	М	-	Molarity, Molar concentration
	n	-	slope of the plot of log p vs. log d
	n ₁	***	slope of the plot of log p vs. log d in LLR
ι,	ⁿ 2		slope of the plot of log p vs. log d in HLR
	n ë	-	calculated slope of the plot of log p vs. log d
	ng		observed slope of the plot of log p vs. log d

: S-5 :

	ⁿ k1	-	slope of the plot of log p vs. log d in LLR for knoop indentations
	ⁿ k2	-	slope of the plot of log p vs. log d in HLR for knoop indentations
	ⁿ v1	••	slope of the plot of log p vs. log d in LLR for vickers indentations
	ⁿ v2		slope of the plot of log p vs. log d in HLR for vickers indentations
	^N 1	-	Polarizer of polarimeter
	^N 2	-	analyzer of polarimeter
	0	-	ordinary ray
	P	-	load in grammes
	p _k	-	pick at kink
v	R	-	radius of a crystal
	RT	-	room temperature
	s		specific rotation
	S	-	diameter of arcs of x-ray powder photograph
	SMS		sodium metasilicate solution
	Sp. gra.	-	specific gravity
	t	-	time

-

: <u>s-</u>6 :

Ŧ	-	absolute temperature
т	-	glass tube of polarimeter
ТА	-	d-tartaric acid solution
$\mathbf{c}^{\mathbf{T}}$		quenching temperature
v	-	function of dissolution speed
Vl		the dissolution velocity (parallel to surface)
٧n	-	the dissolution velocity (normal to surface)
v _{nd}	.	the dissolution velocity at a dislocation site
Vndf	-	the average vertical dissolution velocity of a dislocation free portion of the surface
Vt	-	tangential velocity of dissolution
E	-	compressive strain
9	•••	angle of rotation of optically active substance
0	-	Bragg angle
λ	-	path difference
λ	-	wavelength of x-ray beam

`

v [*] _{h1k111}	-	reciprocal vector corresponding to plane (h ₁ ,k ₁ ,l ₁)
v [*] _{h2^k2¹2}	-	reciprocal vector corresponding to plane (h ₂ ,k ₂ ,l ₂)
S	-	the droplet density
م	****	the density of d-AHP
، م	-	the density of kerosene
5		the compressive stress
б	-	the surface energy per unit area
J		mean free time
Л	-	material quantity